

Atoms- 3.7 Practice Assessment

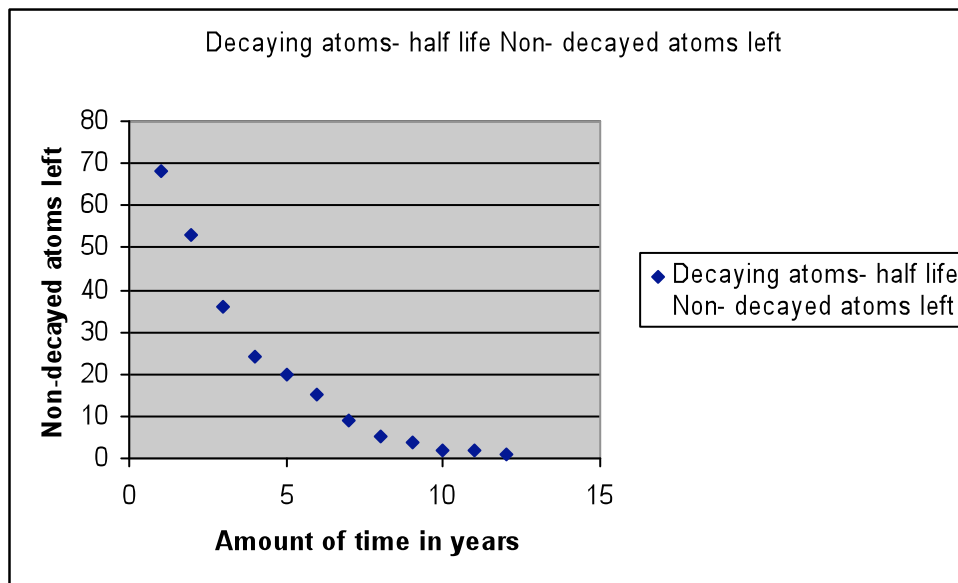
Started with 100 atoms (pins).

Decaying atom

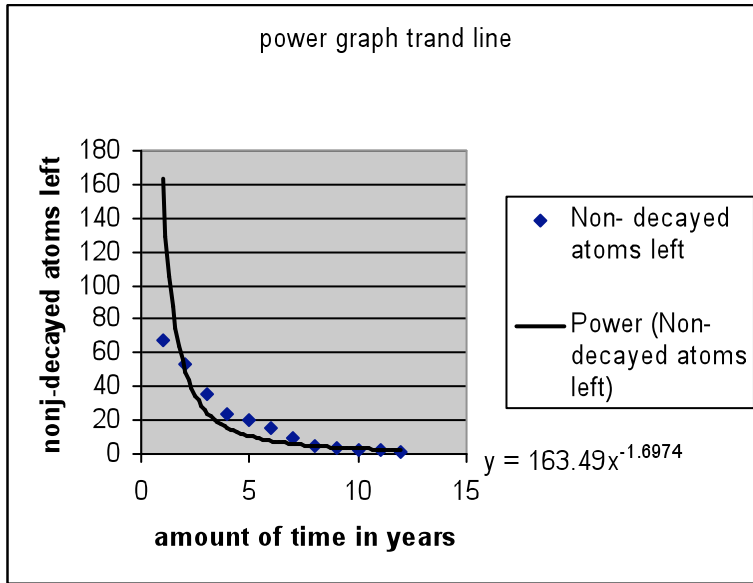
atoms- half life

Amount of time (yrs)	Non- decayed atoms left
1	68
2	53
3	36
4	24
5	20
6	15
7	9
8	5
9	4
10	2
11	2
12	1

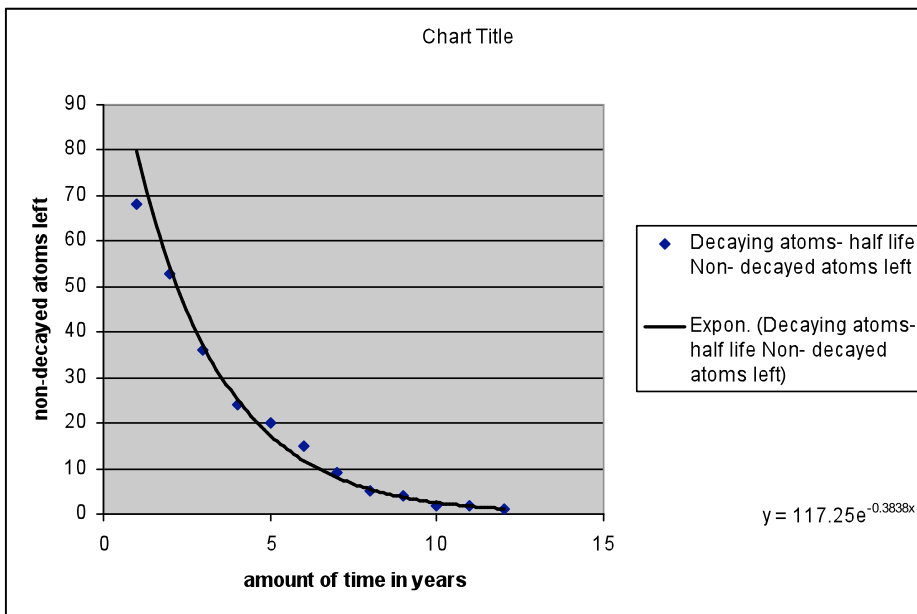
Graphed raw data:



Now to see what kind of graph this is I will use a trend line fitted to the data selecting which one is best fit- power or exponential.



The power trend line is not a very close fit.

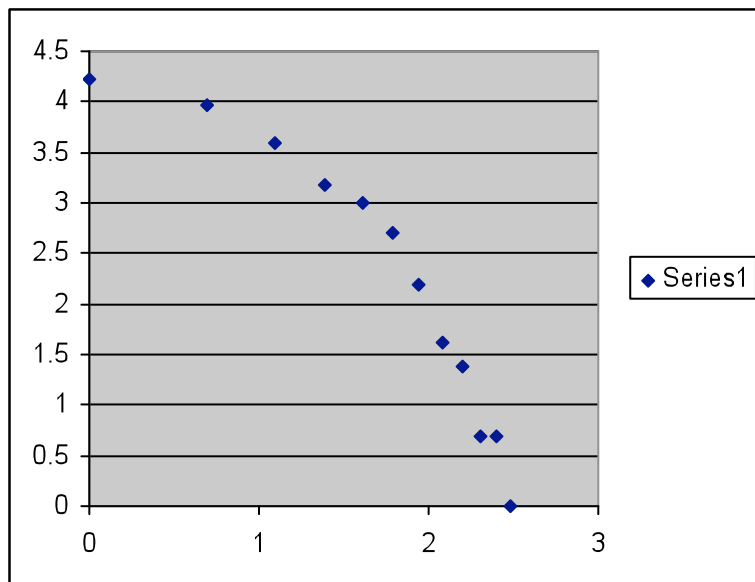


The exponential trend line however is a much closer fit so I can assume that the graph is exponential.

I will check this assumption by logging the data and checking to see if the points worked out produces a straight line.

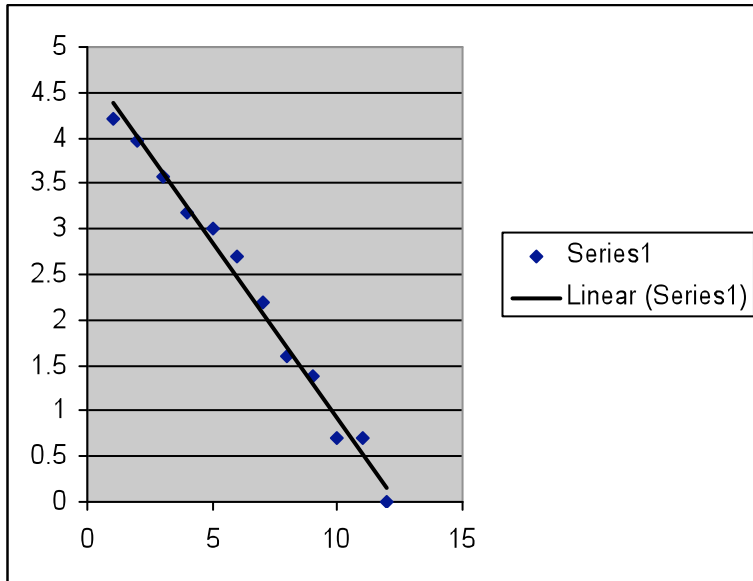
First a power conversion:

Ln (x)	Ln (y)
0	4.219508
0.693147	3.970292
1.098612	3.583519
1.386294	3.178054
1.609438	2.995732
1.791759	2.70805
1.94591	2.197225
2.079442	1.609438
2.197225	1.386294
2.302585	0.693147
2.397895	0.693147
2.484907	0



Then an exponential conversion:

X	ln (y)
1	4.219508
2	3.970292
3	3.583519
4	3.178054
5	2.995732
6	2.70805
7	2.197225
8	1.609438
9	1.386294
10	0.693147
11	0.693147
12	0



The points plotted from the exponential conversion fit the linear graph proving the proving that the graph/equation is exponential.

Graph Equation: $y = 117.25e^{(-0.3838x)}$

Half life of the simulated element:

$$\begin{aligned}
 y &= 117.25e^{(-0.3838x)} \\
 &= 117.25 \times 0.68^x \\
 50 &= 117.25 \times 0.68^x \\
 0.68^x &= 50/117.25 \\
 x \ln 0.68 &= \ln 0.4264 \\
 x &= 2.21 \text{ years}
 \end{aligned}$$